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Review

CHAPTER 9 REVIEW

Stoichiometry SECTION 3

PROBLEMS Write the answer on the line to the left. Show

all your work in the space

provided. 1. 88% The actual

yield of a reaction is 22 g

and the theoretical yield is

25 g. Calculate the

percentage yield. 2. 6.0 mol

of N_2 are mixed with 12.0

mol of H_2 according to the

following equation: $N_2(g)$

$3H_2(g) \rightarrow 2NH_3(g)$? $2NH_3(g)$ N

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Match. Gravity. Created by.
sharknerdo. Terms in this
set (10) Describe the
relationship between

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Avogadro's number and a mole of a substance. A mole is defined as the amount of a substance in 6.022×10^{23} (Avogadro's number) particles.

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when the German airship
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approach to the Naval Air
Station in Lakehurst, New
Jersey. The actual cause of
the explosion is still
unknown, but the entire
volume of hydrogen gas used
to float the airship,

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Stoichiometry Section Review

Problems ? Do section review

problems #1 through #4 on

page 301 of the textbook.

Section Review page 301 1-

The branch of chemistry that

deals with mass

relationships in compounds

and in chemical reactions. 2

HgO ? $2 \text{Hg} + \text{O}_2 \rightarrow 2 \text{HgO}$ a) 2 mol

HgO 2 mol Hg & 2 mol HgO 1

mol O_2 ...

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questions in the space
provided. 1. a.

6 Chemical Bonding
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