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We have observed evidence that when M-NM compounds are dissolved, the metal particles tend to form positively charged ions (cations), while non-metal particles tend to form negatively charged ions (anions).

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by guest predicts a set of

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FORMULA UNIT 1.

Ca<sup>2+</sup> and Br<sup>-</sup> 2. Fe<sup>2+</sup>  
and Cl<sup>-</sup> 3.

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Pd Unit 6 Worksheet 3

Ionic Compounds

Properties Ionic

compounds generally

have higher mp's and

bp's than do molecular

compounds. When



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molten, they conduct electricity. If soluble, the solutions also conduct electricity.

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v2.0 f) Determine

whether substance A

and B will sink or float

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when placed in a bucket of water. A: sink float  
B: sink float (circle correct response)

Defend your answer using the  $m$ - $V$  graph, and your outstanding understanding of density. Refer to the table of

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U6 ws 3 v1.0 3. Next to the formula, give the name of the following simple binary ionic compounds. a.  $\text{Na}_2\text{O}$  b.  $\text{MgCl}_2$  c.  $\text{CaBr}_2$  d.  $\text{Al}_2\text{S}_3$  e.  $\text{CsBr}$  f.  $\text{AgF}$  g.  $\text{Na}_3\text{N}$  h.  $\text{K}_2\text{O}$  5. Write the name of these ionic substances next to the formula, using a Roman numeral to specify the charge of the cation. ...

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develop a model to predict liquid rainout from release of a pressurized, liquefied gas--and, hence the residual amount of material in a vapor cloud, which may be greater than the amount calculated from an enthalpy chart.

RELEASE predicts the rate of fluid discharge, the depressurization,

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flashing and formation of liquid drops, the entrainment of drops into the vapor cloud, the subsequent spreading of the jet, and rate of liquid rainout to a pool on the ground. Designed in a modular fashion to permit adjustment and corrections as new data become available, its multi-layered approach contains sub-models

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Against  
that include the complexities of many variables, including the effect of liquid

superheat, rate of bubble growth, criterion for bubble formation, and heat transfer from the liquid to the growing vapor bubble. To validate RELEASE, CCPS conducted small- and large-scale experiments using



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superheated water, heated liquefied chlorine, methylamine, and cyclohexane that produced valuable data in an area where data are scarce. This book gives complete access, in text and on CD-ROM, to the model and the test data, giving users an informed ability to apply the model to their own

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