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molecules per mole, what air pressure does this represent? Solution:
The mass of one molecule of air may be computed as Molecular weight 28.97 mol – 1

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Solution 1.1. To get started, first list or determine the volumes involved: d = volume of water dumped = 100 cm^3 , c = volume of a sip = 5 cm^3 , and $V = \frac{4}{3} \pi R^2 D$ = volume of water in the oceans = $\frac{4}{3} \pi R^2 D$, where, R is the radius of the earth, D is the mean depth of the oceans, and f is the oceans' coverage fraction. Here we've ignored the ocean volume occupied by salt and have assumed that the ...

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