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of matter that we will
learn about in this
chapter. (If you want
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states of matter, I can refer you to somebody.) Those three states are solid, liquid, and gas. ?

These three states are quite different.

The main difference is in their particles.

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Section 13.4 Changes

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The relationship among the solid, liquid, and vapor states (or phases) of a

substance in a sealed container are best represented in a single graph called a phase diagram Phase diagram- gives the temperature and pressure at which a substances exists as

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solid, liquid, or gas
(vapor) 70.

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13.1 THE NATURE

OF GASES (pages

385–389) This section

introduces the kinetic

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theory and describes how it applies to gases. It defines gas pressure and explains how temperature is related to the kinetic energy of the particles of a substance.

Kinetic Theory and a Model for Gases
(pages 385–386) 1.

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